

## CHEMISTRY OF 2H-ISOINDOLES: RECENT DEVELOPMENTS

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**Abstract.** In this review, new synthetic methods for the preparation of 2H-isoindoles based on ring closure reactions, isoindoline aromatization and ring transformations are discussed. The resulting 2H-isoindoles can be isolated or in situ trapped with dienophiles to give the corresponding Diels-Alder adducts.

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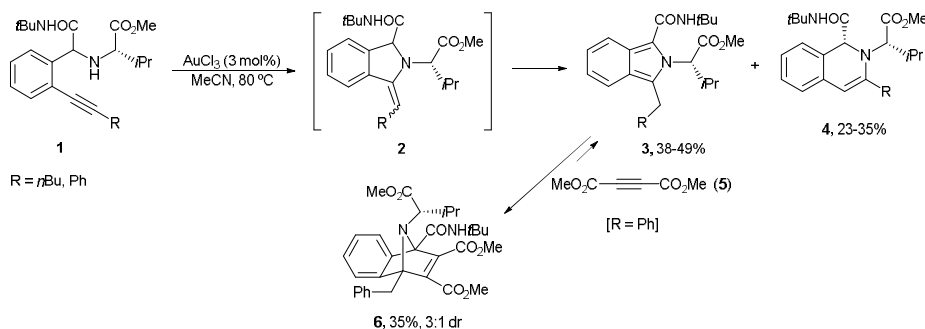
**1. Introduction**

Properties and reactivity of 2H-indoles have attracted considerable attention in the last 50 years.<sup>1–3</sup> The isoindole structure can be found in natural products and bioactive compounds with a pharmacological profile such as antimicrobial, anthelmintic, insecticidal, cyclooxygenase isoenzyme (COX-2), thrombin inhibitor and anticancer activity.<sup>4–6</sup> Isoindole derivatives and oligomers are very important compounds in material science<sup>7–11</sup> and as photosensitizers for photodynamic therapy.<sup>12</sup> Isoindole are rather unstable compounds because of the *o*-quinoid structure and are frequently trapped as latent dienes in intermolecular Diels-Alder (DA) reactions.

Several methodologies for the synthesis of isoindoles have been described in the literature, based mainly on (a) ring-closure reactions, (b) ring transformations, (c) aromatizations, and (d) substituent modifications.<sup>3</sup> In this review article, recent advances in the last 20 years on isoindole chemistry will be considered.

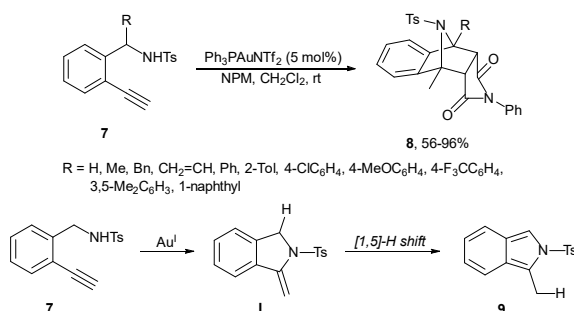
**2. Synthesis by ring-closure reactions****2.1. Alkynes cyclization**

Aromatic compounds bearing an acetylenic unit and nitrogen moieties have been cyclized mainly under transition metal-catalyzed conditions to give different indole derivatives including isoindoles.<sup>13</sup> Intermolecular hydroamination of alkynes **1**, obtained by four-component Ugi reaction, was performed by Dyker and co-workers<sup>14</sup> under AuCl<sub>3</sub> catalysis to give isoindole **3** derived from the 5-*exo-dig*-products **2** by isomerization under the acidic reaction conditions (Scheme 1). Dihydroisoquinolines **4** were also obtained by a 6-*endo-dig* cyclization in 23–35% yield. The chiral isoindole **3a** was allowed to react with dimethyl acetylenedicarboxylate (DMAD) **5** providing cycloadduct **6** in 35% yield as a 3:1 mixture of diastereomers. However, this compound partially decomposed under flash chromatography purification.



**Scheme 1.** Isoindoles **3** from acetylenic compounds **1**.

$\text{Ph}_3\text{PAuNTf}_2$ -catalyzed cycloisomerization-[1,5]-hydride migration-DA reaction of 2-ethynylbenzyl-sulfonamides **7** provided the corresponding isoindole *endo*-cycloadducts **8** by reaction with *N*-phenylmaleimide (NPM) (Scheme 2).<sup>15</sup> This one-pot three-step cascade reaction takes place under very mild reaction conditions in  $\text{CH}_2\text{Cl}_2$  at room temperature (rt). On the basis of deuterium labelling experiments, it has been proposed the initial formation of intermediate **I**, which after [1,5]-hydride transfer gave the isoindole **9**. Various dienophiles such as DMAD **5**, naphthoquinone, tetracyanoethylene and methyleneindolinone were also successfully employed.

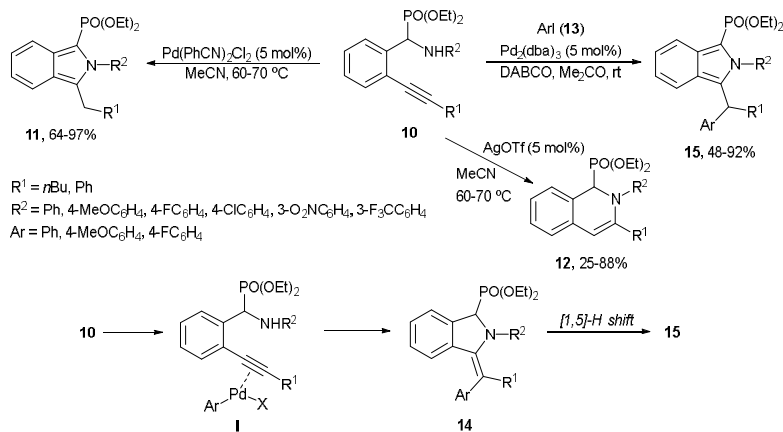


**Scheme 2.** Gold-catalyzed cascade of benzylsulfonamides **7** to isoindole cycloadducts **8**.

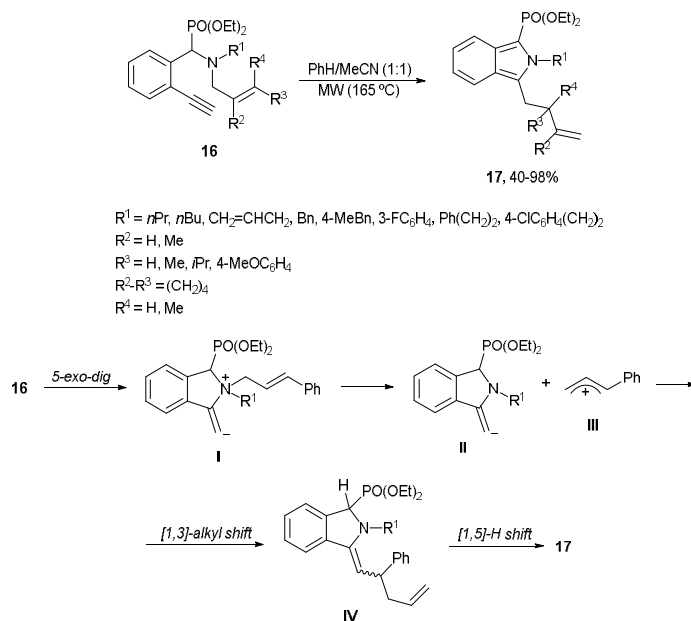
Intramolecular hydroamination of acetylenic  $\alpha$ -amino phosphonates **10**, under Pd-catalysis, provided by a 5-*exo-dig* cyclization isoindoles **11**.<sup>16</sup> Regiodivergent<sup>17</sup> 6-*endo-dig* cyclization to dihydroisoquinolines **12** resulted under AgOTf catalysis. The starting compounds **10** were prepared by  $\text{FeCl}_3$ -catalyzed reaction of 2-alkynylbenzaldehydes with amines and diethyl phosphate.<sup>16</sup> After cyclization of compounds **10** to intermediate **I**, a subsequent [1,5]-H shift took place giving products **11**.<sup>16</sup> On the other hand, in the presence of aryl iodides **13** a [1,3]-aryl shift in intermediate **I** resulted compounds **14**, which after final aromatization provided isoindoles **15** (Scheme 3).<sup>18</sup>

Dieltiens and Stevens<sup>19</sup> reported a metal-free entry to phosphonylated isoindoles **17**, related to compounds **11**, by 5-*exo-dig* cyclization of *N*-allyl amino phosphonates **16** followed by a [1,3]-alkyl shift and final aromatization under microwave (MW) heating (Scheme 4). In this case, it was proposed that after the hydroamination to give intermediate **I**, the [1,3]-alkyl shift produces the anion **II** and cation **III**, which reacted at the phenylated position to yield intermediate **IV** precursor of isoindoles **17**.

Steven's group reported<sup>20</sup> a  $\text{AuCl}_3$ -catalyzed synthesis of 1-cyanoisoindoles **19** from *N*-allylic aminonitriles **18** (Scheme 5). In this case, under MW irradiation no conversion could be detected. In the proposed mechanism, after intramolecular hydroamination promoted by the Lewis acid, intermediate **I** resulted from this 5-*exo-dig* cyclization followed by the [1,3]-alkyl migration to form **II** and 1,5-prototropic aromatization to yield product **19**.



**Scheme 3.** Phosphonylated isoindoles **11** and **15** from acetylenic  $\alpha$ -amino phosphonates **10**.

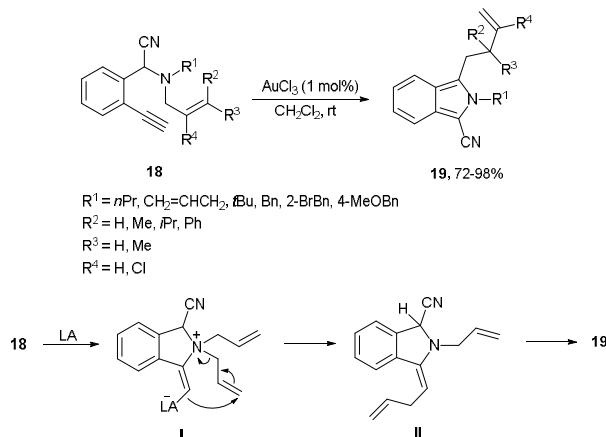


**Scheme 4.** Phosphonylated isoindoles **17** from acetylenic  $\alpha$ -amino phosphonates **16**.

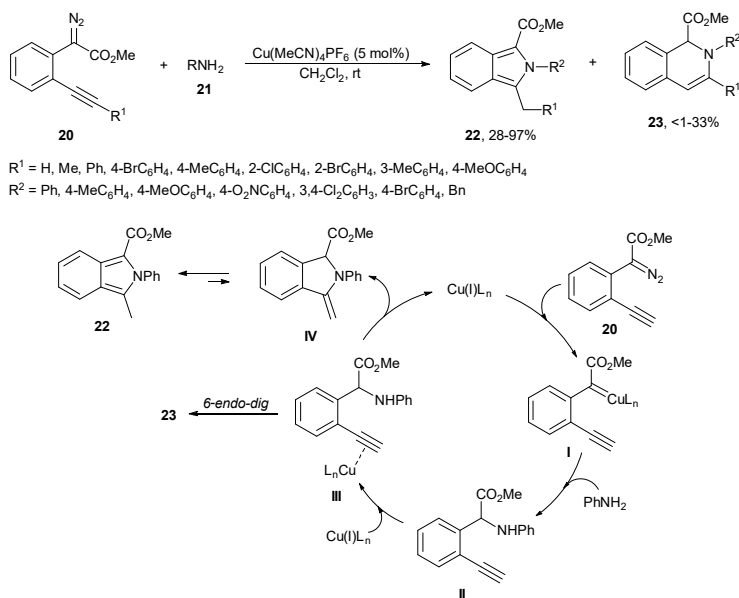
Acetylenic systems bearing a diazoacetate unit at the *ortho*-position **20** underwent metal-catalyzed C–N bond formation through a metal carbene precursor in the presence of primary amines **21** followed by hydroamination generating isoindoles **22** (Scheme 6).<sup>21</sup> This sequential process can be carried out with Cu(MeCN)<sub>4</sub>PF<sub>6</sub> as catalyst under mild reaction conditions. In the proposed mechanism, the Cu(I) carbene **I** is generated from **20**, which reacts with aniline to give intermediate **II**. Subsequently, the amino group attacks the activated triple bond of intermediate **III** in a 5-*exo-dig* manner to provide intermediate **IV** precursor of the isoindole **22**. Cyclization of intermediate **III** in a 6-*endo-dig* manner forms dihydroisoquinolines **23** with yields in the range of <1-33%.

Herndon and co-workers<sup>22</sup> reported the synthesis of naphthalene derivatives **29** through the coupling of electron-deficient alkynes **26**, Fischer carbene complexes **25** and benzaldehyde hydrazones **24** *via* isoindole

intermediates **27** (Scheme 7). The reaction of hydrazones **24** with carbene complexes **25** gave isoindoles **27**, which in the presence of dienophiles afforded azanorbornenes **28** by a regioselective DA reaction. These cycloadducts **28** underwent a nitrene extrusion reaction to form the aromatic products **29** using dioxane as solvent. However, in the presence of a polar solvent such as ethanol, Michael addition products **30** were isolated.



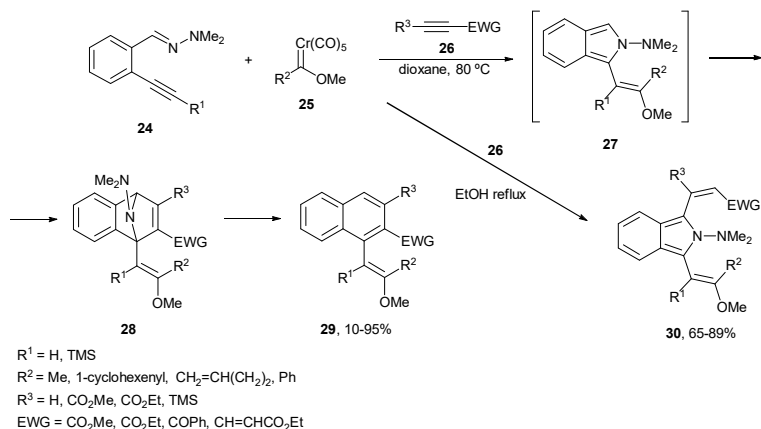
**Scheme 5.** 1-Cyanoisoindoles **19** from acetylenic aminonitriles **18**.



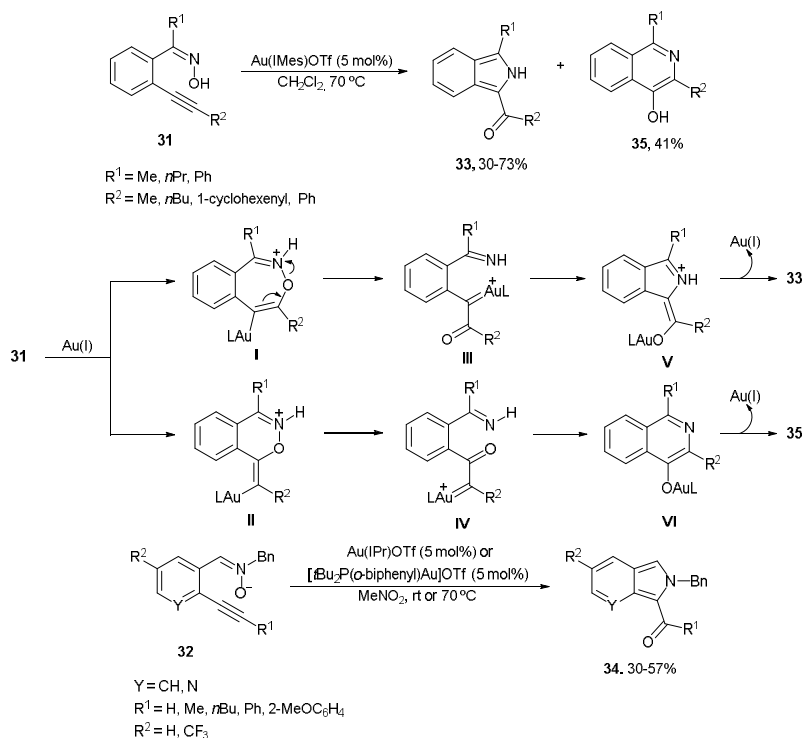
**Scheme 6.** Isoindoles **22** from acetylenic phenyldiazoacetates **20**.

Acetylenic *Z*-ketoximes **31** or nitrones **32** underwent a gold-catalyzed redox cascade cyclization leading to isoindoles **33** and **34**, respectively (Scheme 8).<sup>23</sup> In the presence of [1,3-bis-(2,4,6-trimethylphenyl)imidazol-2-ylidene]Au[OTf] [Au(Imes)OTf], *o*-alkynylaryl ketoximes **31** in dichloromethane at 70 °C provided the corresponding isoindoles **33** in good yields, except in the case of **31** ( $\text{R}^1=\text{Me}$ ,  $\text{R}^2=\text{Ph}$  and  $\text{R}^1=\text{R}^2=\text{Ph}$ ), which gave 4-hydroxyisoquinolines **35** as well. These results were

rationalized based on the formation of intermediate **I** or **II** through a 7-*endo-dig* or 6-*exo-dig* cyclization, respectively, followed by a redox process to give the Au-carbenoid **III** or **IV**, respectively, giving the five and six-membered products by intermediacy of **V** and **VI**, respectively. Nitrones **32** experimented cyclization using either [1,3-bis-(2,6-diisopropyl)imidazol-2-ylidene]Au]OTf [Au(IPr)OTf] or [tBu<sub>2</sub>P(*o*-biphenyl)Au]OTf in MeNO<sub>2</sub> giving exclusively isoindoles **34** in moderate yields by a similar mechanism than Z-oximes.



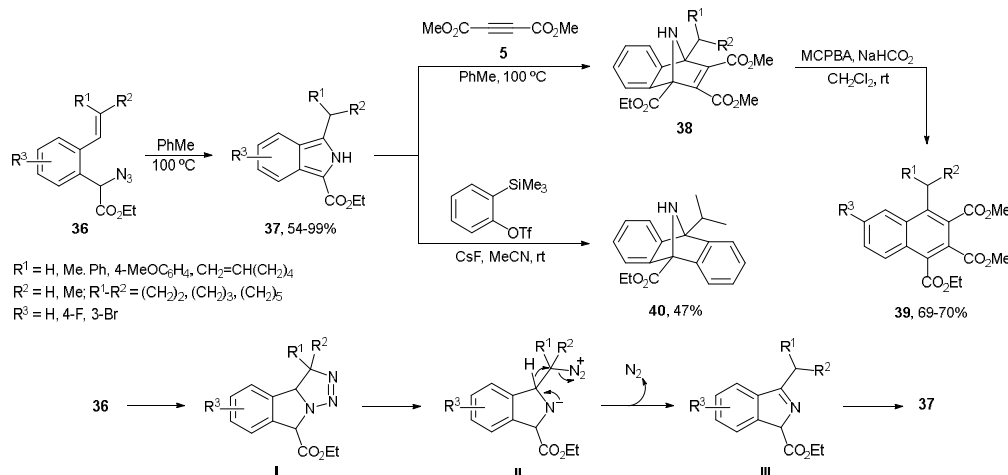
**Scheme 7.** Synthesis of naphthalene derivatives **29** by intermediacy of isoindoles **27**.



**Scheme 8.** Isoindoles **33** and **34** from Z-oximes **31** and nitrones **32**, respectively.

## 2.2. 1,3-Dipolar cycloadditions

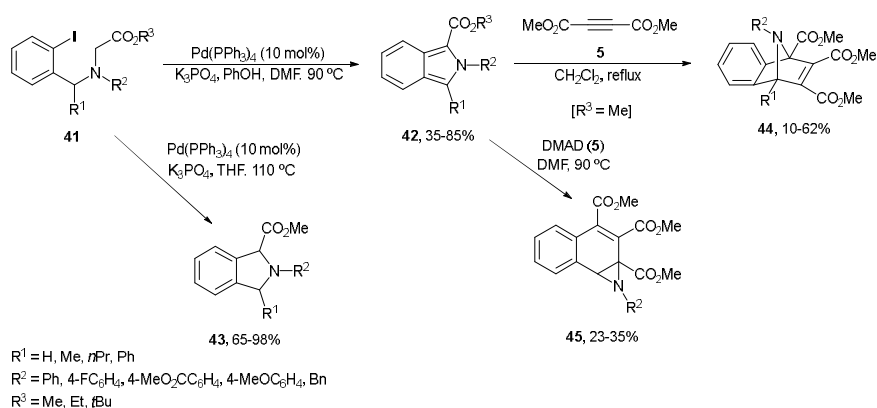
$\alpha$ -Azido carbonyl compounds bearing a 2-alkenylaryl moiety at the *ortho*-position **36** gave under thermal conditions isoindoles **37** (related to compounds **22** in Scheme 6) *via* 1,3-dipolar cycloaddition of azides onto alkenes (Scheme 9).<sup>24,25</sup> This process took place in toluene at 100 °C either starting from azides **36** or by reaction of the mesylate precursor of **36** with NaN<sub>3</sub> in dimethylformamide (DMF) to form the crude azides **36**. In the proposed mechanism, the formation of triazoline **I** followed by subsequent elimination of nitrogen in intermediate **II** and [1,5]-H shift in intermediate **III** gave product **37**. The obtained isoindoles **37** were allowed to react with DMAD **5** to give cycloadducts **38**, which were transformed into 1,2,3,4-tetrasubstituted naphthalenes **39** after final oxidative deamination in very good overall yields. The cycloaddition with benzyne took place at room temperature affording cycloadduct **40** in 47% yield.



Scheme 9. Synthesis of naphthalenes **39** and bicycle **40** from isoindoles **37**.

## 2.3. Intramolecular $\alpha$ -arylation of $\alpha$ -amino esters

Solé and Serrano<sup>26,27</sup> described the synthesis of 1-isoindolecarboxylic acid esters **42** by Pd-catalyzed intramolecular arylation of  $\alpha$ -(2-iodobenzylamino) esters **41** (Scheme 10). Using Pd(PPh<sub>3</sub>)<sub>4</sub> with K<sub>3</sub>PO<sub>4</sub> and a catalytic amount of phenol in DMF at 90 °C, isoindoles **42** were formed, whereas working without phenol in THF at 110 °C isoindolines **43** were obtained.<sup>27</sup>

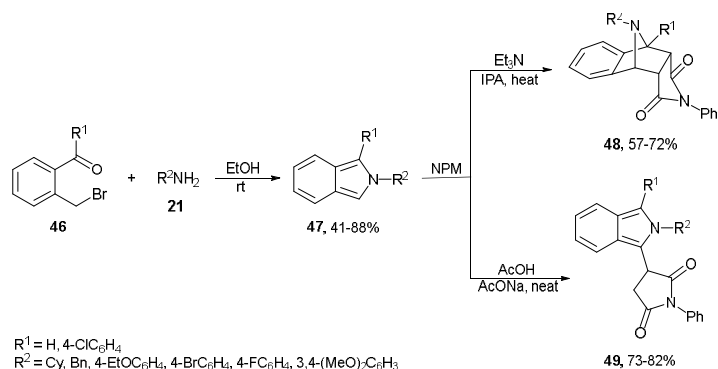


Scheme 10. 1-Isoindolecarboxylic acid esters **42** from  $\alpha$ -(2-iodobenzylamino) esters **41**.

When isoindole derivatives **42** were allowed to react with DMAD **5**, the corresponding adducts **44** were mainly obtained in dichloromethane, but also aziridines **45** were prepared working in DMF.<sup>26</sup>

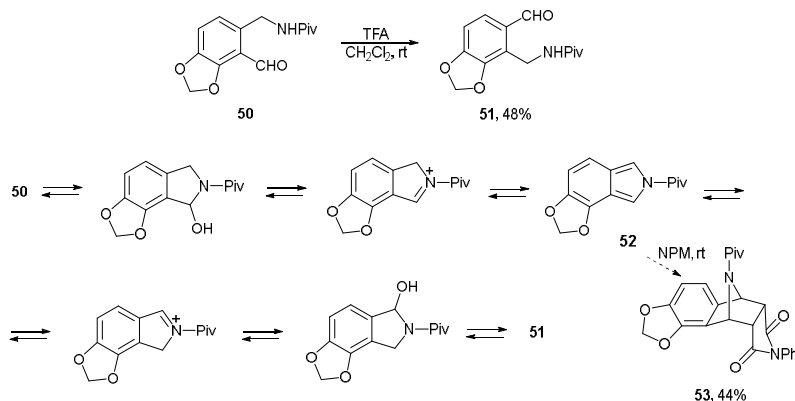
#### 2.4. Cyclization of benzylamines

1-Substituted isoindoles **47** have been prepared from 2-(bromomethyl)benzaldehyde or benzophenone **46** and primary amines **21** in EtOH at room temperature (Scheme 11).<sup>28</sup> Under isopropanol reflux and in the presence of Et<sub>3</sub>N isoindoles **47** reacted with NPM to give cycloadducts **48**. However, under AcOH reflux and NaOAc as base Michael adducts **49** were obtained.



**Scheme 11.** 1-Substituted isoindoles **47** from 2-(bromomethyl)benzaldehyde or benzophenone **46** and DA reactions.

Under acidic conditions *o*-(pivaloylaminomethyl)benzaldehyde substituted by methylenedioxy group **50** underwent a rearrangement reaction leading to the regioisomers **51** (Scheme 12).<sup>29</sup> The proposed mechanism for this transformation involves the formation of an isoindole **52** as key intermediate, which was trapped with NPM to give cycloadduct **53**.



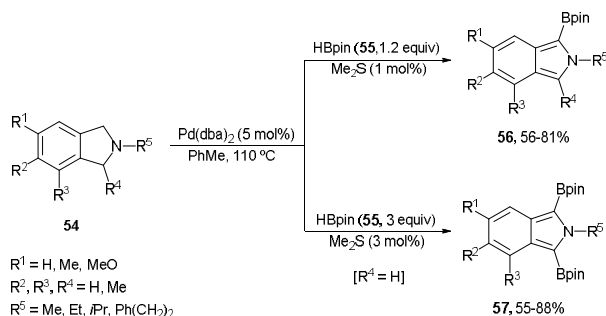
**Scheme 12.** Rearrangement of *o*-(pivaloylaminomethyl)benzaldehyde **50** to **51** through intermediacy of isoindole **52**.

#### 3. Synthesis by aromatization processes

A general method for the synthesis of isoindoles is the aromatization of isoindolines.<sup>3,30</sup> These saturated heterocycles can be prepared by different cyclization strategies such as (a) amination of dihalides, (b) intramolecular hydroamination, (c) DA reactions, (d) [2+2+2] cyclotrimerization of alkynes and (e) cyclocarbonylative Sonogashira reaction, which have been recently reviewed.<sup>30</sup>

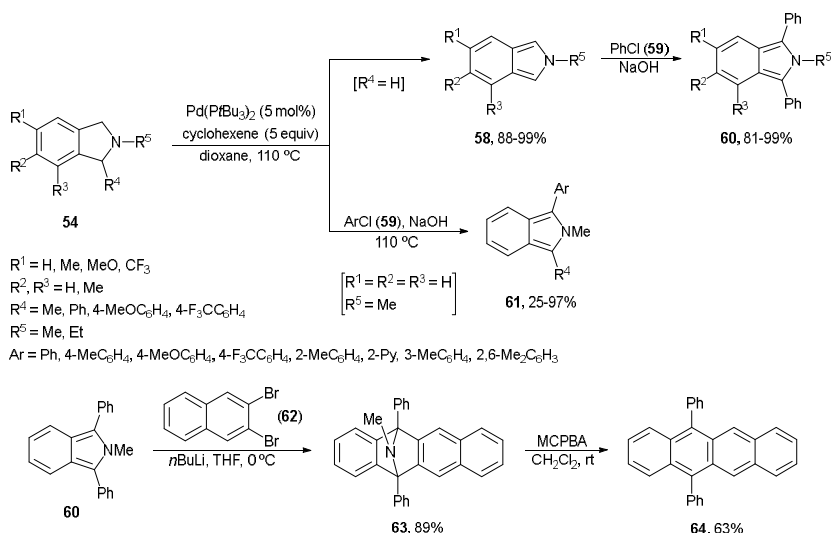
### 3.1. C–H functionalization of isoindolines

1-Borylisoindoles **56** were prepared *via* Pd-catalyzed dehydrogenation/C–H borylation of isoindolines **54** (Scheme 13).<sup>31</sup> The reaction of *N*-substituted isoindolines **54** with 1.2 equivalents of pinacolborane (HBpin) **55** took place first by dehydrogenation to isoindole followed by C–H borylation, which was accelerated by the presence of Me<sub>2</sub>S. The second borylation was also achieved working with three equivalents of HBpin **55** starting from isoindolines **54** (R<sup>4</sup>=H) to provide doubly borylated isoindoles **57**. These borylated isoindoles can be submitted to Suzuki–Miyaura coupling with aryl iodides.



**Scheme 13.** Borylated isoindoles **56** and **57** from isoindolines **54**.

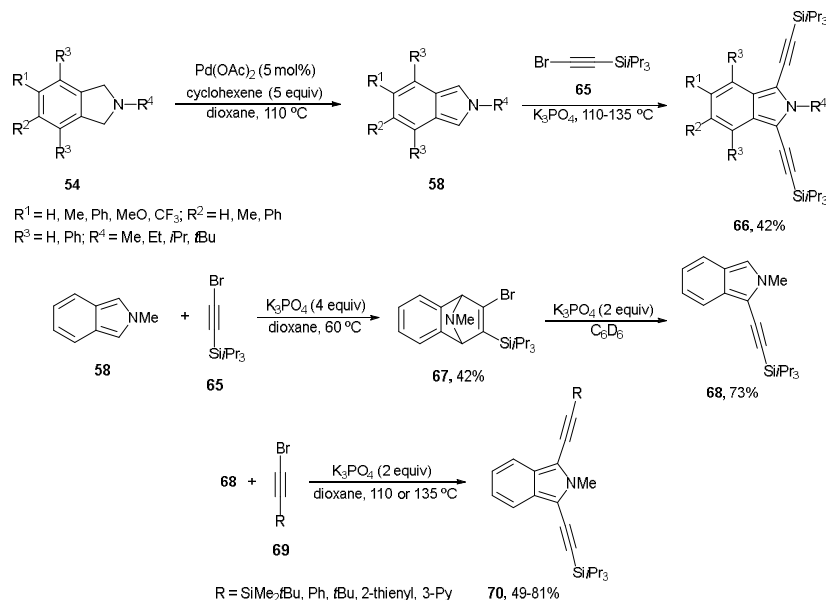
Aryl substituted isoindoles have been obtained by a dehydrogenative/C–H arylation process starting from isoindolines **54** performed by the same group (Scheme 14).<sup>32</sup> In this case, the first dehydrogenation step allowed to isolate isoindoles **58** or to perform *in situ* the C–H arylation step adding phenyl chloride **59** to provide diarylated isoindoles **60**. When monosubstituted isoindolines were used, the one-pot dehydrogenation/C–H arylation with aryl chlorides **59** gave *N*-methyl isoindoles **61**. DA reaction of *N*-methyl 1,3-diphenylisoindole **60** with naphthalene generated *in situ* from 2,3-dibromonaphthalene **62**, afforded bicyclic amine **63**, which was converted into 5,12-diphenyltetracene **64** by treatment with *meta*-chloroperbenzoic acid.



**Scheme 14.** Arylated isoindoles **60** and **61** from isoindolines **54**.



The Suginome's group<sup>33</sup> has carried out the synthesis of 1,3-dialkylated isoindoles **66** by reaction of isoindoles **58** with (bromoethyl)triisopropylsilane **65** (Scheme 15). In the absence of transition metal, first the [4+2] cycloaddition takes place first to give **67**, followed by ring opening with elimination of HBr to yield the monoalkylated isoindole **68** without formation of a cycloadduct. Unsymmetrical 1,3-dialkynylisoindoles **70** can be prepared *via* alkynylation of monoalkylated isoindole **68** with bromoalkynes **69**. A one-pot synthesis of isoindoles **66** was carried out starting from isoindolines in good yields. The triisopropyl group in compound **66** was desilylated with tetra-*n*-butylammonium fluoride and also submitted to desilylative Sonogashira coupling with aryl iodides in the presence of AgF. Photophysical properties of alkynylated isoindoles showed strong fluorescence of compounds **66**.



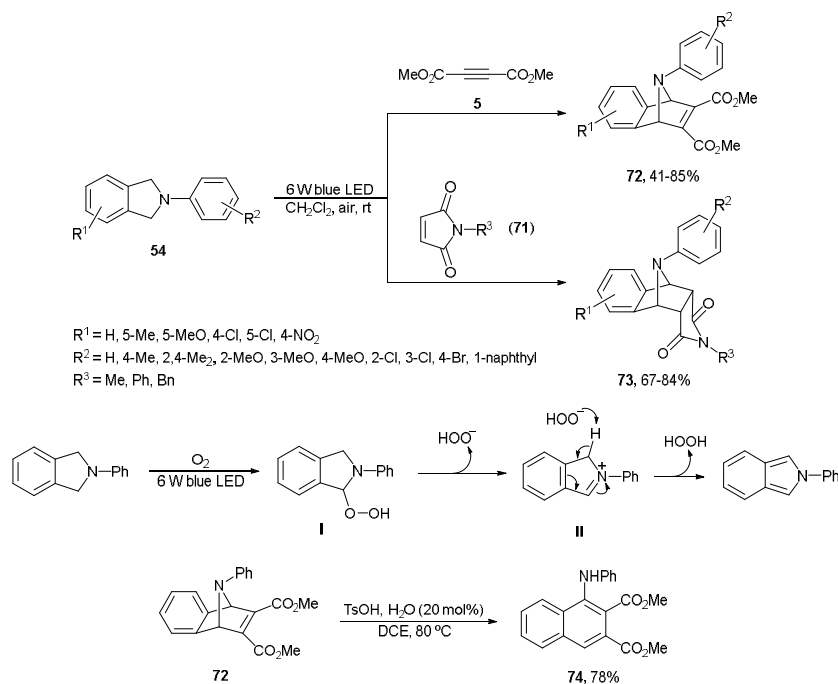
**Scheme 15.** Alkynylated isoindoles **66**, **68** and **70** from isoindolines **54**.

Transition metal-free isoindole formation from isoindolines in the presence of air has been carried out using visible-light which enables an intermolecular DA reaction under mild reaction conditions.<sup>34</sup> Thus, the reaction of isoindolines **54** with DMAD **5** or maleimides **71** took place with 6 W blue LED in dichloromethane at room temperature to give products **72** and **73** in excellent diastereoselectivity and high yields (Scheme 16). A plausible mechanism for forming *N*-phenylisoindole from *N*-phenylisoindoline involves the formation of hydroperoxide **I** in the presence of oxygen. Elimination of hydrogen peroxide ion from **I** provides iminium **II** which isomerizes to *N*-phenylisoindole. Treatment of cycloadduct **72** ( $R^1=R^2=H$ ) with *p*-TsOH afforded 1-naphthylamine derivative **74** in 78% yield.

### 3.2. [1,5]-Hydride shift

Gold-catalyzed [1,5]-H shift of *N*-propargylisoindolines **75** and subsequent DA reaction with dienophiles has been reported by Gong and co-workers.<sup>35</sup> The *in situ* generated *N*-allylisoindoles such as **76** gave cycloadducts **77**, **79**, **81**, **83** and **84** in good yields and diastereoselectivities working in 1,2-dichloroethane (DCE) at 60 °C (Scheme 17). This two-step procedure gave *endo*-cycloadducts **77** with maleimides **71** and products **79** with dimethyl fumarate **78**. However, dimethyl maleate **80** provides product **81** as a 1:1 mixture of diastereomers. Benzylidene malononitrile **82** showed less reactivity giving **83** in 52% yield and DMAD **5** reacted with isoindoles **76** after cooling down the temperature to 20 °C providing products **84**. On the basis of deuterium labelling experiments a plausible mechanism showed in Scheme 17 was proposed. The [1,5]-H shift gives intermediate **I**, which undergoes a reversible protonation and

deprotonation sequence under the assistance of  $\text{TF}_2\text{N}^-$  forming the  $\sigma$ -gold complex **II** and finally the *N*-allylisoindole **76**.



**Scheme 16.** Isoindoles by visible light oxidation of isoindolines **54** and *in situ* DA reaction.

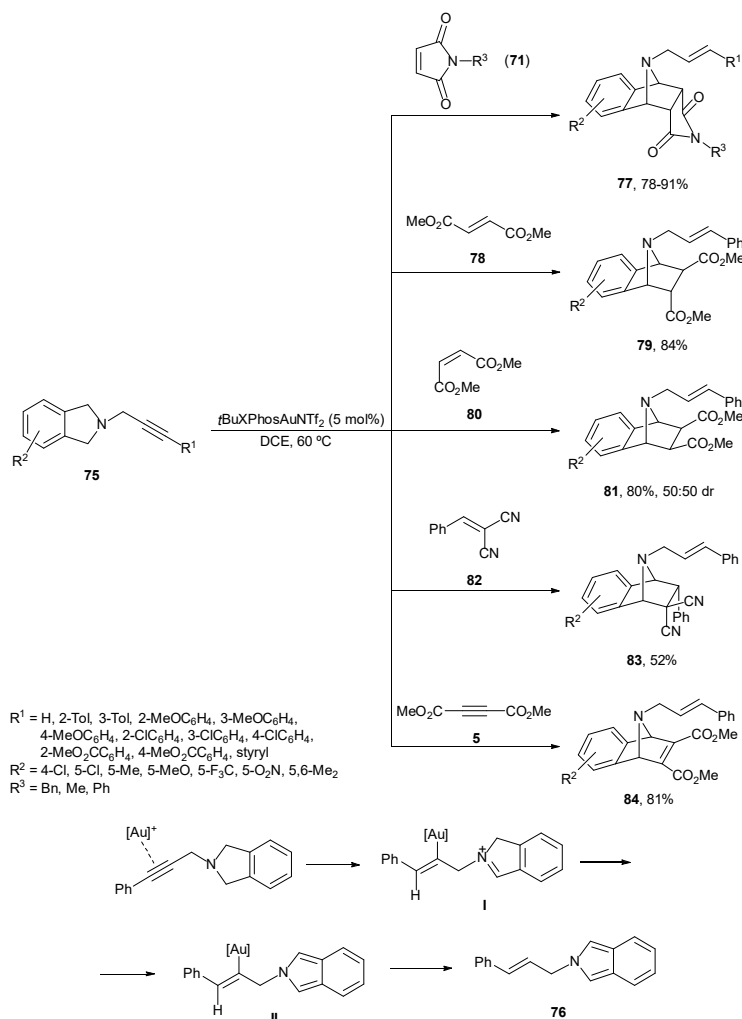
Diphenyl phosphoric acid (DPP) **86** promoted the [1,5]-H shift of isoindolines **85** to provide isoindoles **87** which were also trapped by DMAD **5** and maleimides **71** giving cycloadducts **88** and **89**, respectively (Scheme 18).<sup>35</sup> In the proposed mechanism, after protonation of the benzylic alcohol, deprotonation takes place to give intermediate **I**. Then [1,5]-H shift process occurs providing the iminium cation intermediate **II**, which undergoes deprotonation to isoindole **87**. On the other hand, when the more acidic *p*-TsOH was used instead of DPP **86** in the presence of DMAD **5** as dienophile, 1-naphthylamine derivative **90** was obtained by protonation of the cycloadduct **88** and isomerization.

The Cu(I)-catalyzed three component-coupling of an alkyne, an aldehyde and an amine<sup>36</sup> has been applied to the synthesis of isoindole **95** by Ma and co-workers.<sup>37</sup> The coupling of isoindoline **91** with cyclohexanecarbaldehyde **92** and 1,1-dimethylpropargyl alcohol **93** gave isoindole **95** by an unexpected Cu(I)-catalyzed *E*-stereoselective reduction of the *in situ* formed propargyl amine **94** via [1,5]-H shift (Scheme 19). The corresponding isoindole **95** was trapped by addition of *N*-methylmaleimide (NMM) at -10 °C to provide cycloadduct **96** in 70% overall yield.

#### 4. Synthesis by ring transformation

Retro DA reactions have been used for the synthesis of moderate stable isoindoles. Swager and co-workers<sup>38</sup> reported the synthesis of 4,5,6,7-tetrafluoroisoindole **99** by reaction of azabicyclo **97** with 3,6-(di-2-pyridyl)-1,2,3,4-tetrazine **98** at room temperature (Scheme 20), adapted from the Warrenner's methodology<sup>39</sup> described by Gribble.<sup>40</sup> This process takes place *via* a DA reaction and subsequent thermally allowed electrocyclic fragmentation.<sup>40</sup> This isoindole **99** was transformed into tetracenes **102** by DA reaction with naphthalynes resulting from dibromonaphthalenes **100** to give products **101**, which were aromatized by deamination with an aqueous solution of NaOH to furnish products **102** in 15–35% overall yield. These fluorinated tetracenes **102** have good solubility in common organic solvents and crystallize with antiparallel,

cofacial, stacked column superstructures and **102** ( $R^1=H$ ,  $R^2=n\text{-C}_8\text{H}_{17}$ ) exhibited a slipped parallel  $\pi$ -stacking arrangement in crystals. They are promising candidates for the use in organic electronics.

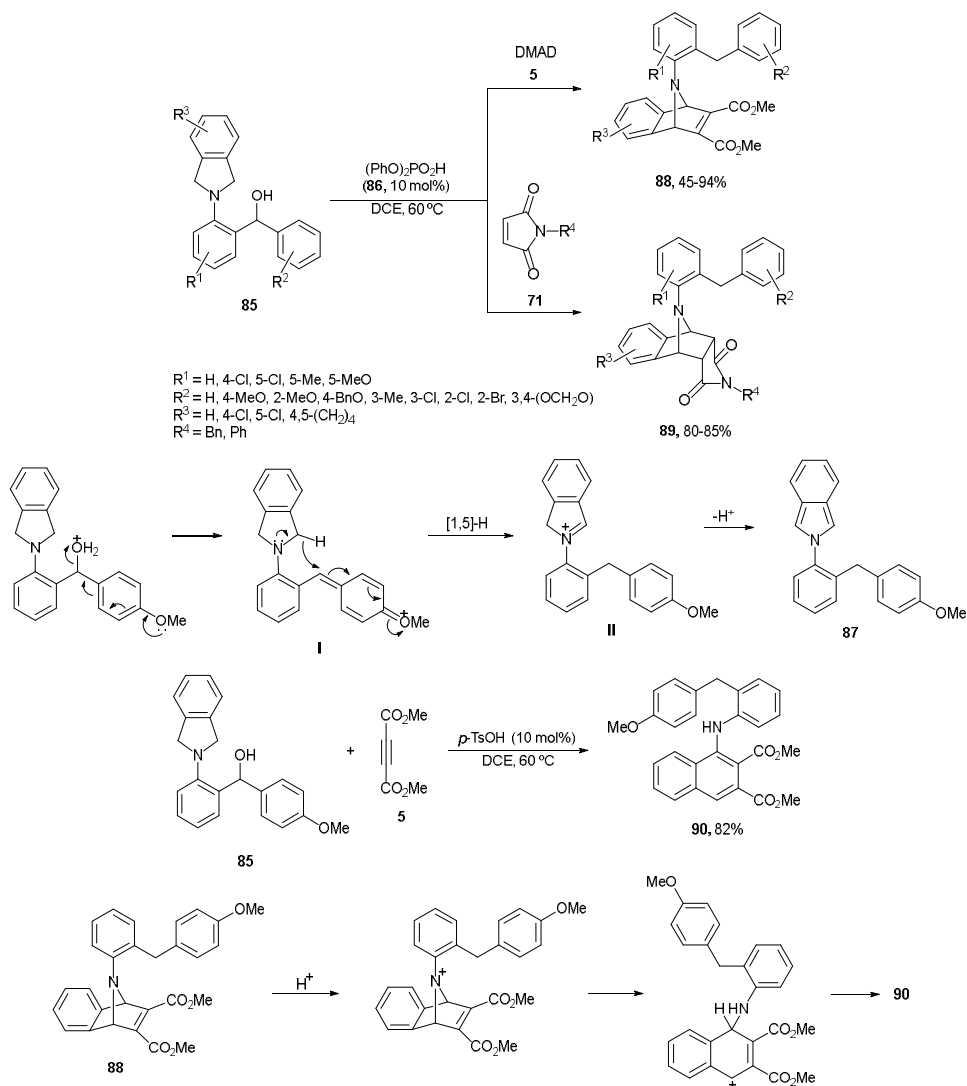


**Scheme 17.** *N*-Allylic isoindoles **76** from isoindolines **75** trapped in DA reactions.

Oxadisilole-fused isoindoles **105** and **106** have been prepared using Warrenner's methodology starting from azabicycles **103** and **104**, respectively, by Lee and co-workers (Scheme 21).<sup>41</sup> The resulting isoindoles **105** and **106** were allowed to react with different dienophiles such as DMAD **5** and benzyne giving cycloadducts in good yields (48-98%). Some representative examples have been submitted to a deamination process by treatment of benzyne cycloadduct **107** with trifluoroacetic acid at room temperature to provide *p*-quinone **108**. In the case of cycloadduct **109** additional treatment with AcOH at 100 °C gave quinone **110** in 45% yield.

Rincón and Plumet<sup>42</sup> applied the Warrenner's methodology to the preparation of *N*-Boc-protected isoindole **112** using the azabicycle **111** and the tetrazine **98** (Scheme 22). This isoindole **112** was trapped by different alkynyl and vinyl sulfones **113** to provide cycloadducts **114-116** in moderate to good yields.

Bicyclopyrroles **117** have been transformed into the corresponding stable isoindoles **118** performing the retro-DA reaction using supercritical carbon dioxide avoiding oxidation of isoindoles when high-boiling solvents were used (Scheme 23).<sup>43</sup> Oxidative decomposition of isoindoles was prevented by adding ethylene gas as an oxygen scavenger increasing the isolation yield of **118** ( $R=X=H$ ) with 68–81% yields.

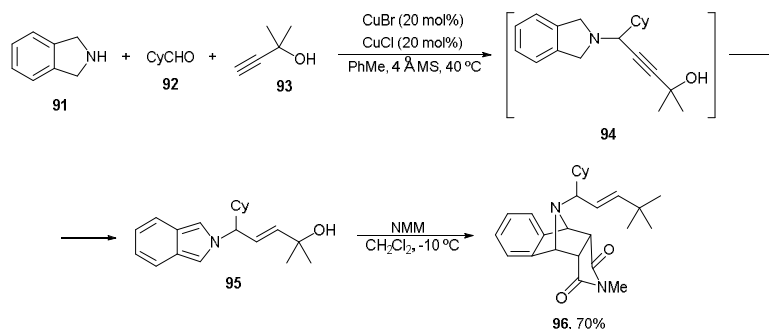


**Scheme 18.** Isoindoles **87** from isoindolines **85** trapped in DA reactions.

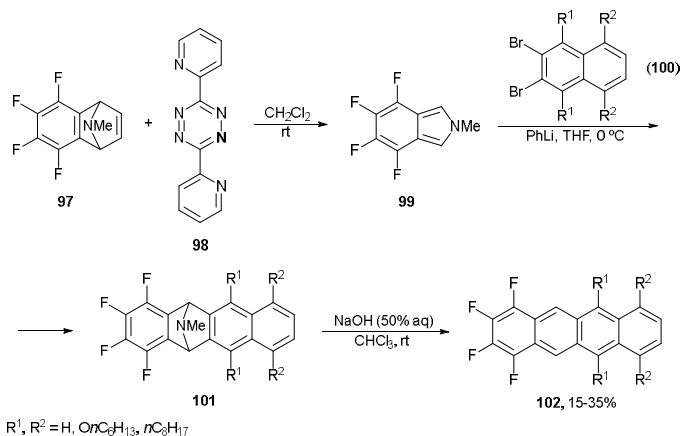
## 5. Other methodologies

Recent advances in isoindole chemistry are related to the transformation of nucleophilic isoindoles into electrophilic isoindolinium *via* protonation of the *in situ* generated isoindoles. Wang and co-workers<sup>44</sup> reported the reaction of 2-(bromomethyl)benzaldehydes **46** with triptamines **119** to give isoindoles **120**, which were treated *in situ* with trifluoroacetic acid giving isoindolines **121** by a Pictet-Spengler-type cyclization in good yields (Scheme 24).

A Pd-catalyzed Heck-type dearomative [4+2] annulation of isoindoles **122** with internal alkynes **123** provided polycyclic pyrrolidines derivatives **124** in good yields (Scheme 25).<sup>45</sup> This process has been carried out with *N*-(2-bromo)- and also with (2-iodo)aryl isoindoles **122**. In the proposed mechanism, after the initial oxidative addition of the aryl iodide to Pd(0), the arylpalladium species **I** is formed. Then, the insertion of the internal alkyne provides an alkenylpalladium intermediate **II** followed by an intramolecular Heck reaction with the carbon-carbon double bond closed up to the ester group of the isoindole delivering intermediate **III**. The subsequent aromatization produces the benzylic palladium species **IV**, which undergoes  $\beta$ -hydride elimination to the final product and the catalytic species is regenerated by  $K_2CO_3$ .



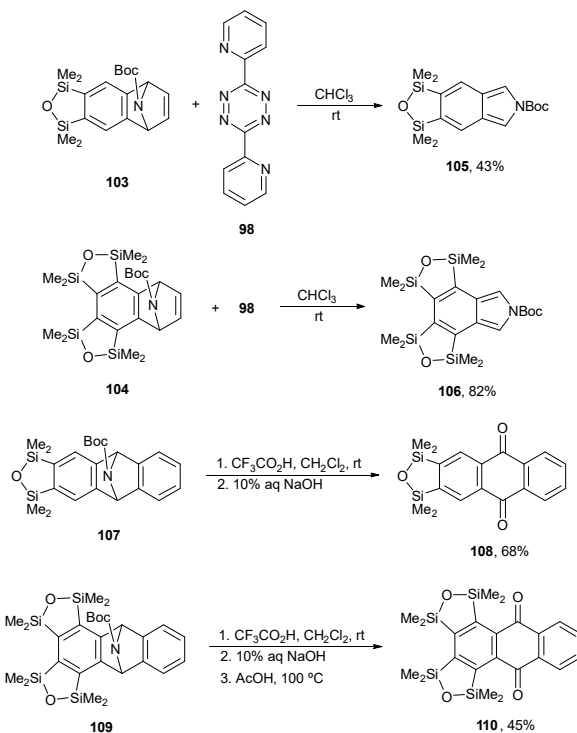
**Scheme 19.** Isoindole **95** formed by a three component-coupling of isoindoline **91**, cyclohexanecarbaldehyde **92** and 1,1-dimethylpropargyl alcohol **93**.



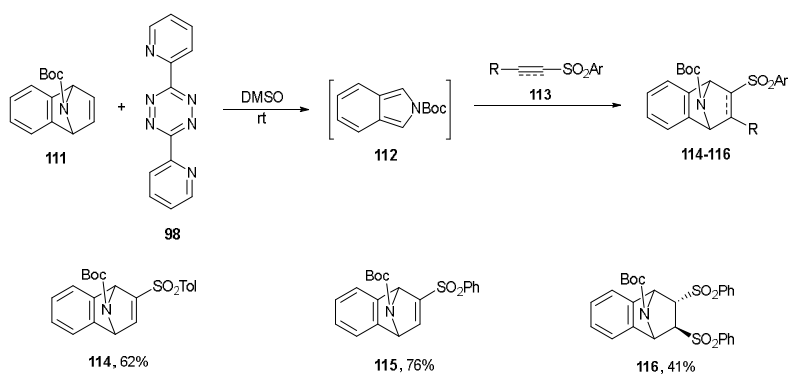
**Scheme 20.** 4,5,6,7-Tetrafluoroisoindole **99** from azabicyclic **97** and tetrazine **98**.

## 6. Conclusion

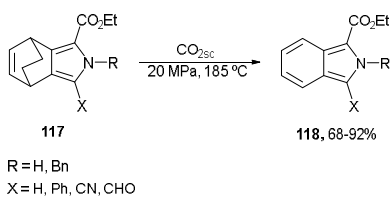
In the present review, recent developments for the synthesis of 2*H*-indoles indicate that four strategies have been mainly employed. Intramolecular cyclization of *o*-alkynyl benzylamine derivatives under Au, Pd and also under metal-free conditions are important methodologies based on hydroamination reactions followed by aromatization. Acetylenic diazoacetates reacting with amines under Cu-catalysis to give similar derivatives are alternative starting compounds. Aromatization process starting from isoindolines and [1,5]-hydrogen shift of *N*-propargyl isoindolines are mediated by Lewis or Brønsted acids. Concerning ring transformation Warrener's methodology is still an interesting strategy. Other methodologies allowed the intramolecular  $\alpha$ -alkylation of isoindoles bearing a tryptamine and by an intermolecular Heck-type [4+2] annulation. In many cases it is possible to isolate relative stable isoindoles and alternatively have been trapped by a DA reaction.



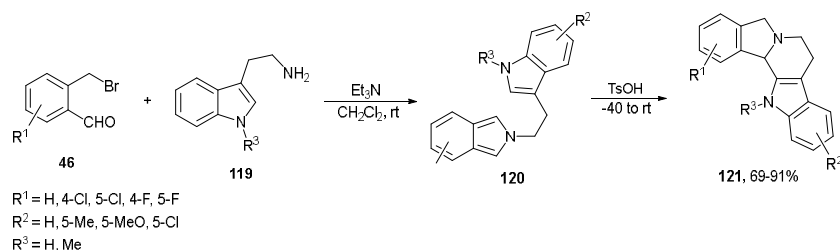
**Scheme 21.** Ozadisilole-fused isoindoles **105** and **106** by Warrenner's method.



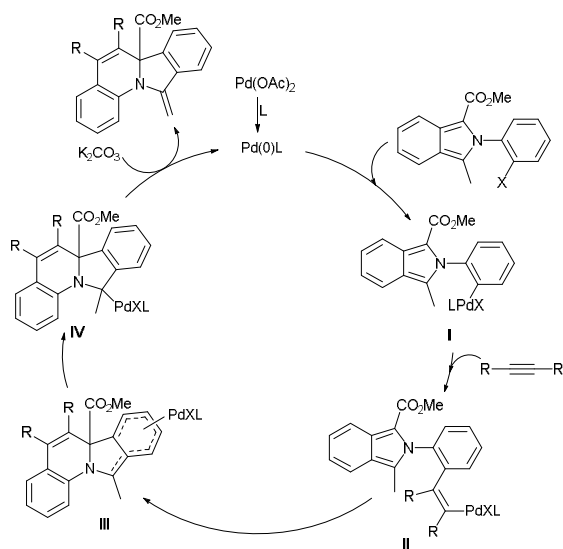
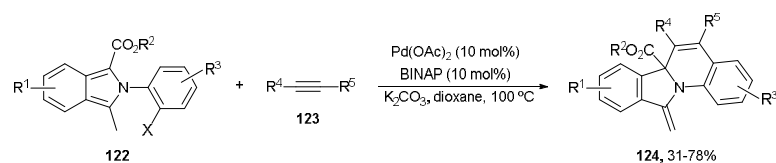
**Scheme 22.** *N*-Boc-isoindole **112** by Warrenner's method.



**Scheme 23.** Isoindoles **118** by retro-DA reaction of bicyclopyrroles **117** in supercritical  $\text{CO}_2$ .



**Scheme 24.** Isoindolines **121** by intramolecular 1-alkylation of isoindoles **120**.



**Scheme 25.** Dearomative annulation of isoindoles **122** with alkynes **123** under Pd-catalysis.

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## References

- White, T. L. *Diels-Alder Chemistry with Isoindole and Isobenzofuran*; University of California: Santa Barbara, CA, 1994.
- Jones, G. B.; Chapman, B. J. In *Comprehensive Heterocyclic Chemistry II*; Katritzky, A. R.; Rees, C. W.; Scriven, E. F. V., Eds.; Pergamon: Oxford, UK, 1996.
- Donohoe, T. J. In *Fused Five Membered Heteroarenes with One Heteroatom*; Thomas, E. J., Ed.; Science of Synthesis; Georg Thieme Verlag: Stuttgart, 2000.
- Subbarayappa, A.; Patoliya, P. U. *Indian J. Chem., Sect. B: Org. Chem. Incl. Med. Chem.* **2009**, *48*, 545-552.
- Speck, K.; Magauer, T. *Beilstein J. Org. Chem.* **2013**, *9*, 2048-2078.
- Bathia, B. K. *Curr. Topics Med. Chem.* **2017**, *17*, 189-207.
- Liégault, B.; Lapointe, D.; Caron, L.; Vlassova, A.; Fagnou, K. *J. Org. Chem.* **2009**, *74*, 1826-1834.
- Campeau, L.-C.; Stuart, D. R.; Leclerc, J.-P.; Bertrand-Laperle, M.; Villemure, E.; Sun, H.-Y.; Laserre, S.; Guimond, N.; Lecavallier, M.; Fagnou, K. *J. Am. Chem. Soc.* **2009**, *131*, 3291-3306.
- Jafarpour, F.; Rahiminejadan, S.; Hazrati, H. *J. Org. Chem.* **2010**, *75*, 3109-3112.
- Varotto, A.; Nam, C.-Y.; Radivojevic, I.; Tomé, J. P. C.; Cavaleiro, J. A. S.; Black, C. T.; Drain, C. M. *J. Am. Chem. Soc.* **2010**, *132*, 2552-2554.
- Mori, S.; Nagata, M.; Nakahata, Y.; Kazumasa, Y.; Goto, R.; Mutsumi, K.; Taya, M. *J. Am. Chem. Soc.* **2010**, *132*, 4054-4055.
- Hammerer, F.; Achelle, S.; Baldeck, P.; Maillard, P.; Teulade-Fichou, M.-P. *J. Phys. Chem.* **2011**, *115*, 6503-6508.
- Neto, J. S. S.; Zeni, G. *Asian J. Org. Chem.* **2021**, *10*, 1282-1320.
- Kadzymirsz, D.; Hildebrandt, D.; Merz, K.; Dyker, G. *Chem. Commun.* **2006**, *2006*, 661-662.
- Zhang, S.; Cheng, B.; Wang, S.-A.; Zhou, L.; Tung, C.-H.; Wang, J.; Xu, Z. *Org. Lett.* **2017**, *19*, 1072-1075.
- Ding, Q.; Ye, Y.; Fan, R.; Wu, J. *J. Org. Chem.* **2007**, *72*, 5439-5442.
- Nájera, C.; Beletskaya, I. P.; Yus, M. *Chem. Soc. Rev.* **2019**, *48*, 4515-4618.
- Ding, Q.; Wang, B.; Wu, J. *Tetrahedron Lett.* **2007**, *48*, 8599-8602.
- Dieltsiens, N.; Stevens, C. V. *Org. Lett.* **2007**, *9*, 465-468.
- Heugebaert, T. S. A.; Stevens, C. V. *Org. Lett.* **2009**, *11*, 5018-5021.
- Peng, C.; Cheng, J.; Wang, J. *Adv. Synth. Catal.* **2008**, *350*, 2359-2364.
- Duang, S.; Sinha-Mahapatra, D. K.; Herndon, J. W. *Org. Lett.* **2008**, *10*, 1541-1544.
- Yeom, H.-S.; Lee, Y.; Lee, J.-E.; Shin, S. *Org. Biomol. Chem.* **2009**, *7*, 4744-4752.
- Hui, B. W.-Q.; Chioba, S. *Org. Lett.* **2009**, *11*, 729-732.
- Tong, B. M. K.; Hui, B. W.-Q.; Chua, S. H.; Chiba, S. *Synthesis* **2011**, *2011*, 3552-3562.
- Solé, D.; Serrano, O. *Org. Biomol. Chem.* **2009**, *7*, 3382-3384.
- Solé, D.; Serrano, O. *J. Org. Chem.* **2010**, *75*, 6267-6270.
- Voitenko, Z. V.; Sypchenko, V. V.; Levkov, I. V.; Potikha, L. M.; Kovtunencko, V. A.; Shishkin, O. V.; Shishkina, S. V. *J. Chem. Res.* **2011**, *2011*, 615-618.
- Hargitai, C.; Koványi-Lax, G.; Nagy, T.; Abrányi-Balogh, P.; Dancsó, A.; Halász, J.; Tóth, G.; Simig, G.; Volk, B. *Monatsh. Chem.* **2019**, *150*, 1121-1125.
- Albano, G.; Aronica, L. A. *Synthesis* **2018**, *50*, 1209-1227.
- Ohmura, T.; Kijima, A.; Suginoma, M. *J. Am. Chem. Soc.* **2009**, *131*, 6070-6071.
- Ohmura, T.; Kijima, A.; Suginome, M. *Org. Lett.* **2011**, *13*, 1238-1241.
- Ohmura, T.; Kijima, A.; Komori, Y.; Suginome, M. *Org. Lett.* **2013**, *15*, 3510-3513.
- Liu, C.; Zhen, L.; Cheng, Y.; Du, H.-J.; Zhao, H.; Wen, X.; Kong, L.-Y.; Xu, Q.-L.; Sun, H. *Org. Lett.* **2015**, *17*, 2684-2687.
- Zhen, L.; Dai, L.; Yu, S.-Q.; Lin, C.; Sun, H.; Xu, Q.-L. *Eur. J. Org. Chem.* **2017**, *2017*, 560-569.
- Peshkov, V. A.; Pereshivko, O. P.; Van der Eycken, E. V. *Chem. Soc. Rev.* **2012**, *41*, 3790-3807.
- Fan, W.; Yuan, W.; Ma, S. *Nat. Commun.* **2014**, *5*, 3884-3892.
- Chen, Z.; Müller, P.; Swager, T. M. *Org. Lett.* **2006**, *8*, 273-276.
- Priestley, G. M.; Warrenner, R. N. *Tetrahedron Lett.* **1972**, *13*, 4245-4298.



40. Lettoulhier, C. S.; Gribble, G. W. *J. Org. Chem.* **1983**, *48*, 2364-2366.
41. Chen, Y.-L.; Lee, M.-H.; Wong, W.-Y.; Lee, A. W. M. *Synlett* **2006**, *2006*, 2510-2512.
42. Rincón, R.; Plumet, J. *Synlett* **2008**, *2008*, 911-913.
43. Ito, S.; Akaki, M.; Shinozaki, Y.; Iwabe, Y.; Furuya, M.; Tobata, M.; Roppongi, M.; Sato, T.; Itoh, N.; Oba, T. *Tetrahedron Lett.* **2017**, *38*, 1338-1342.
44. Weintraub, R. A.; He, W.; Wang, X. *Tetrahedron Lett.* **2020**, *61*, 152128.
45. Yao, T.; Zhang, F.; Zhang, J.; Liu, L. *Org. Lett.* **2020**, *23*, 5063-5067.